

## ДОПОЛНИТЕЛЬНЫЕ МАТЕРИАЛЫ К СТАТЬЕ

**МЕДЬМАРГАНЦЕВЫЙ КАТАЛИЗАТОР ДЛЯ НИЗКОТЕМПЕРАТУРНОГО ОКИСЛЕНИЯ СО НА ОСНОВЕ СМЕШАННОГО ОКСИДА CuMnO<sub>2</sub>**

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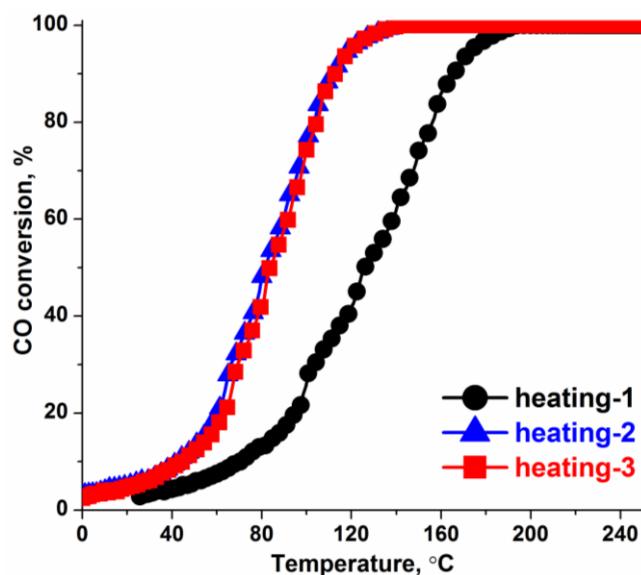
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**Table S1. Structural characteristics (cell parameters ( $a$ ,  $c$ ), volume of lattice cell ( $V$ ), crystallite size ( $D$ )) of additional to CuMnO<sub>2</sub> phases in the composition of studied copper-manganese catalysts**

Sample	Mn <sub>3</sub> O <sub>4</sub>	CuO	Cu <sub>2</sub> O	Cu <sub>y</sub> Mn <sub>3-y</sub> O <sub>4</sub>
CuMn-1	T=80°C	-	D=23 nm	-
	T=120°C	D=22 nm	-	-
	T=180°C	D=23 nm	D=55 nm	-
CuMn-2	pH=9.6	a=5.761(1) Å, c=9.429(2) Å V=313.0(1) Å <sup>3</sup> , D=11 nm	D=14 nm	D=17 nm
	pH=11.8	a=5.760(2) Å, c=9.439(5) Å V=313.1(1) Å <sup>3</sup> , D=17 nm	D=17 nm	D=33 nm
	pH=12.4	a=5.770(4) Å, c=9.47(1) Å V=315.2(1) Å <sup>3</sup> , D=11 nm	-	-
CuMn-3	CO+O <sub>2</sub> , 350°C	-	-	-
				a=8.282(1) Å, D=8 nm



**Fig.S1 Catalytic curves of CO conversion measured during the 1<sup>st</sup>, 2<sup>nd</sup>, and 3<sup>rd</sup> heating of CuMn-1 (T=80°C) sample in the reaction mixture of 0.24%CO/1%O<sub>2</sub>/Ar**

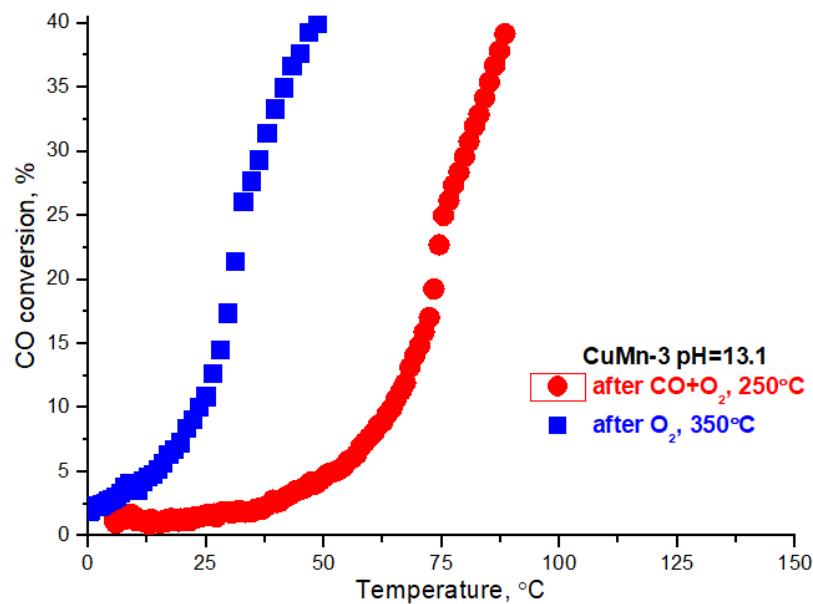


Fig.S2 Catalytic curves of CO conversion measured during the 3<sup>rd</sup> heating of CuMn-3 ( $pH=13.1$ ) sample: as-prepared and pretreated in 20%  $O_2/Ar$  at  $350^\circ C$  for 1 h

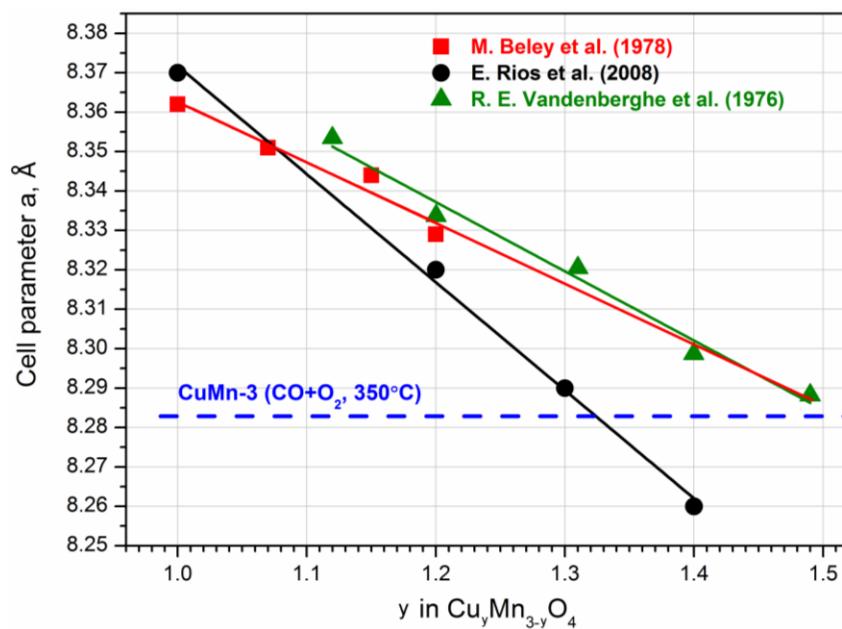


Fig.S3 The lattice parameter of  $Cu_yMn_{3-y}O_4$  spinel versus  $y$  value

Table S2. Quantitative spectral characteristics of copper and manganese for spent CuMn-2 samples

Sample		Mn 2p <sub>3/2</sub> , eV	$\Delta E_1$ , eV ( $\pm 0.2$ )	Cu 2p <sub>3/2</sub> , eV	Cu <sup>+</sup> , %	Cu/Mn
CuMn-2	pH=9.6	641.4	10.2	933.6	10	0.88
	pH=12.4	641.4	10.5	932.3	42	0.93
	pH=13.1	641.4	10.2	932.3	43	0.86