

CRYSTAL STRUCTURE, MICROHARDNESS
AND THERMAL EXPANSION OF TERNARY TaIr₂B₂ BORIDEV.V. Lozanov¹, A.V. Utkin¹, G.A. Letyagin², G.V. Romanenko², D.M. Polyukhov²,
S.G. Kozlova^{3,4}, A.T. Titov⁵, N.I. Baklanova¹¹*Institute of Solid State Chemistry and Mechanochemistry SB RAS, Novosibirsk, Russia*
E-mail: lozanov.25@yandex.ru²*International Tomography Center SB RAS, Novosibirsk, Russia*³*Nikolaev Institute of Inorganic Chemistry SB RAS, Novosibirsk, Russia*⁴*Belgorod Shukhov State Technological University, Belgorod, Russia*⁵*Sobolev Institute of Geology and Mineralogy SB RAS, Novosibirsk, Russia*Received
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The ternary TaIr₂B₂ boride was prepared by the reaction between iridium and TaB₂ powders. It was shown that a choice in favor of any crystal structure of TaIr₂B₂ boride using only X-ray powder diffraction analysis and single-crystal XRD is rather difficult to make. The DFT calculations were carried out to predict the energetically preferred crystal structure of TaIr₂B₂. The ¹¹B solid-state NMR spectrum of TaIr₂B₂ was recorded for the first time. A combination of diffraction and spectroscopic methods, as well as theoretical calculations, allowed us to make clear choice in favor of the triclinic space group $P\bar{1}$ for the crystal structure of TaIr₂B₂. The Vickers microhardness value for TaIr₂B₂ was found to be 17.9±2.3 GPa, and the ternary TaIr₂B₂ boride can be considered a hard material. Therefore, the family of hard ternary boride phases got a new member. The coefficients of thermal expansion of TaIr₂B₂ measured by *in situ* high-temperature XRD analysis are independent of temperature along the *a* and *b* axes, but the CTE along the *c* axis deviates noticeably at elevated temperatures. The findings for the new ternary TaIr₂B₂ boride are of interest for the rational design of complex structural materials containing Ta—Ir—B-based components and prediction of their high-temperature behavior. Further research into this ternary boride as well as other iridium-containing ternary borides will provide a tool to solve the problems faced by high-temperature materials science.

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EDN: AYSYPS**Key words:** ternary boride, iridium, crystal structure, thermal expansion, microhardness.

INTRODUCTION

Transition metal borides have a rather wide range of applications resulting from their refractoriness, hardness, superconductivity, as well as mechanical, magnetic, and optical properties [1]. A variety of properties inherent to transition metal borides are related to the fact that they have a rich crystal chemistry and can form crystal structures ranging from 1D to 3D networks.

The family of ternary borides are characterized by an even greater diversity of «structural–chemical–physical» relations. Among ternary borides, those simultaneously containing early and late transition metals (in particular, platinum-group metals) are of special interest.

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CONCLUSIONS

The ternary TaIr_2B_2 boride was prepared by the reaction between iridium and tantalum diboride powders at elevated temperatures. The products were comprehensively characterized by a set of modern analytical techniques, including X-ray powder diffraction and single-crystal XRD, SEM/EDX at different accelerating voltages, DFT calculations, and ^{11}B solid-state MAS NMR spectroscopy. It was shown that a clear choice in favor of the Cc , $C2/c$ or $P\bar{1}$ crystal structure of ternary TaIr_2B_2 boride using only the common diffraction methods, including X-ray powder diffraction and single-crystal XRD, is rather difficult to make. The DFT calculations showed that the $P\bar{1}$ and $C2/c$ crystal structures for TaIr_2B_2 compound are energetically more preferred (~ 10 eV) than the Cc crystal structure, but poorly distinguishable from each other ($\Delta E \sim 0.02$ eV). Additional solid-state NMR studies were conducted to solve this problem. The ^{11}B solid-state NMR spectrum of ternary TaIr_2B_2 boride was recorded for the first time. The combination of diffraction, theoretical and spectroscopic methods allowed us to make a clear choice in favor of the triclinic space group $P\bar{1}$ for the crystal structure of the ternary TaIr_2B_2 phase.

The Vickers microhardness value for the TaIr_2B_2 ternary phase was measured to be 17.9 ± 2.3 GPa. Based on this result, ternary TaIr_2B_2 boride can be considered a hard material. This value is also comparable to those previously reported for the ternary Hf—Ir—B phases. Therefore, a new member has been added to the family of hard ternary boride phases. The coefficients of thermal expansion were measured for TaIr_2B_2 immediately by *in situ* high-temperature XRD analysis. The CTEs of TaIr_2B_2 along the a and b axes are independent of temperature and equal to $7.3 \cdot 10^{-6} \text{ K}^{-1}$ and $7.2 \cdot 10^{-6} \text{ K}^{-1}$, respectively. The temperature dependence of the CTE along the c axis noticeably deviates at elevated temperatures. The CTE of the new TaIr_2B_2 ternary boride is close to the values determined by us for hafnium–iridium ternary boride.

Therefore, the findings on the crystal structure, microhardness and thermal expansion of new ternary TaIr_2B_2 boride are of interest for rational design of the complex structural materials containing Ta—Ir—B-based components and prediction of their high-temperature behavior. There are grounds for hope that further research into this ternary boride and other ternary borides containing iridium will provide a tool to solve problems faced by the high-temperature materials science.

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Supporting information. The Supporting Information is available: Synthesis, Single crystal data, NMR spectra, Powder XRD patterns and EDX analysis, Thermal expansion of the ternary TaIr_2B_2 boride (PDF).

REFERENCES

1. G. Akopov, M.T. Yeung, R.B. Kaner. Rediscovering the crystal chemistry of borides. *Adv. Mater.*, **2017**, 29, 1604506. <https://doi.org/10.1002/adma.201604506>
2. N.I. Baklanova, V.V. Lozanov, A.T. Titov. The first evidence of the high oxidation resistance of the novel ternary tantalum-iridium-boron phase. *Corros. Sci.*, **2019**, 160, 108178. <https://doi.org/10.1016/j.corsci.2019.108178>
3. V.V. Lozanov, A.V. Utkin, T.A. Gavrilova, A.T. Titov, A.I. Beskrovny, G.A. Letyagin, G.V. Romanenko, N.I. Baklanova. New hard ternary Hf—Ir—B borides formed by reaction hafnium diboride with iridium. *J. Am. Ceram. Soc.*, **2022**, 105(3), 2323–2333. <https://doi.org/10.1111/jace.18234>

4. V.V. Lozanov, A.V. Utkin, A.A. Vasin, M.A. Sheindlin, N.I. Baklanova. Hot press assisted synthesis and thermophysical properties of iridium intermetallic compounds. *Thermochim. Acta*, **2020**, 689, 178641. <https://doi.org/10.1016/j.tca.2020.178641>
5. V.V. Lozanov, N.I. Baklanova, N.V. Bulina, A.T. Titov. New ablation-resistant material candidate for hypersonic applications: Synthesis, composition, and oxidation resistance of HfIr₃-based solid solution. *ACS Appl. Mater. Interfaces*, **2018**, 10(15), 13062–13072. <https://doi.org/10.1021/acsami.8b01418>
6. N.I. Baklanova, V.V. Lozanov, A.T. Titov. One-step preparation of TaIr₃-based material and its ablation performance under extreme environmental conditions. *Corros. Sci.* **2018**, 143, 337–346. <https://doi.org/10.1016/j.corsci.2018.08.044>
7. K. Górnicka, X. Gui, B. Wiendlocha, L.T. Nguyen, W. Xie, R.J. Cava, T. Klimczuk. NbIr₂B₂ and TaIr₂B₂ – New low symmetry noncentrosymmetric superconductors with strong spin–orbit coupling. *Adv. Funct. Mater.*, **2021**, 31(3), 2007960. <https://doi.org/10.1002/adfm.202007960>
8. A.M. Barrios Jiménez, U. Burkhardt, R. Cardoso-Gil, K. Höfer, S.G. Altendorf, R. Schlögl, Y. Grin, I. Antonyshyn. Hf₂B₂Ir₅: a self-optimizing catalyst for the oxygen evolution reaction. *ACS Appl. Energy Mater.*, **2020**, 3(11), 11042–11052. <https://doi.org/10.1021/acsaeam.0c02022>
9. O. Sichevych, S. Flipo, A. Ormeci, M. Bobnar, L. Akselrud, Y. Prots, U. Burkhardt, R. Gumeniuk, A. Leithe-Jasper, Y. Grin. Crystal structure and physical properties of the cage compound Hf₂B_{2–28}Ir_{5+δ}. *Inorg. Chem.*, **2020**, 59(19), 14280–14289 <https://doi.org/10.1021/acs.inorgchem.0c02073>
10. Q. Zheng, R. Gumeniuk, H. Borrmann, U. Burkhardt, W. Schnelle, A. Leithe-Jasper, Y. Grin. TM₇TM'₆B₈ (TM = Ta, Nb; TM' = Ru, Rh, Ir): New AlB₂-related compounds. In: Abstract Book: 17th Int. Symp. Boron, Borides Relat. Mater., İstanbul, Türkiye, Sept 11–17, 2011 / Eds. O. Yücel, L. Özmen. İstanbul, Turkey: İSBB O.C., **2011**, 201.
11. Q. Zheng, M. Kohout, R. Gumeniuk, N. Abramchuk, H. Borrmann, Y. Prots, U. Burkhardt, W. Schnelle, L. Akselrud, H. Gu, A. Leithe-Jasper, Y. Grin. TM₇TM'₆B₈ (TM = Ta, Nb; TM' = Ru, Rh, Ir): new compounds with [B₆] ring polyanions. *Inorg. Chem.*, **2012**, 51(14), 7472–7483. <https://doi.org/10.1021/ic201978n>
12. E.M. Carnicom, W. Xie, T. Klimczuk, J. Lin, K. Górnicka, Z. Sobczak, N.P. Ong, R.J. Cava. TaRh₃B₂ and NbRh₃B₂: superconductors with a chiral noncentrosymmetric crystal structure. *Sci. Adv.*, **2018**, 4(5), eaar 7969. <https://doi.org/10.1126/sciadv.aar7969>
13. C. Benndorf, M. de Oliveira, C. Doerenkamp, F. Haarmann, T. Fickenscher, J. Kösters, H. Eckert, R. Pöttgen. ¹¹B and ⁸⁹Y solid state MAS NMR spectroscopic investigations of the layered borides YTB₄ (T = Mo, W, Re). *Dalton Trans.*, **2019**, 48 (3), 1118–1128. <https://doi.org/10.1039/C8DT04444A>
14. Y. Kishimoto, Y. Kawasaki, T. Ohno. ¹¹B NMR study of magnetism in RRh₃B₂ (R = La, Ce, Nd, Sm, Eu and Gd). *J. Phys. Soc. Jpn.*, **2004**, 73(7), 1970–1981. <https://doi.org/10.1143/JPSJ.73.1970>
15. J. Roger, V. Babizhetskyy, S. Cordier, J. Bauer, K. Hiebl, L. Le Pollès, S. Elisabeth Ashbrook, J.-F. Halet, R. Guérin. Crystal structures, physical properties and NMR experiments on the ternary rare-earth metal silicide boride compounds RE₅Si₂B₈ (RE = Y, Sm, Gd, Tb, Dy, Ho). *J. Solid State Chem.*, **2005**, 178 (6), 1851–1863. <https://doi.org/10.1016/j.jssc.2005.02.023>
16. I. Zeiringer, X. Cheng, X.-Q. Chen, E. Bauer, G. Giester, P.F. Rogl. Crystal structures and constitution of the binary system iridium-boron. *Sci. China Mater.*, **2015**, 58(8), 649–668. <https://doi.org/10.1007/s40843-015-0078-6>
17. V.V. Lozanov, N.I. Baklanova. Vliyanie geksaborida kal'tsiya na fazoobrazovanie v sistemakh diborid gafniya–iridii i diborid tantala–iridii (The effect of calcium hexaboride on the phase formation in the hafnium diboride–iridium and tantalum diboride–iridium systems). *Tr. Kol'sk. Nauchn. Tsentra Ross. Akad. Nauk*, **2023**, 14(4), 15–20. <https://doi.org/10.37614/2949-1215.2023.14.4.002> (In Russ.)
18. G.M. Sheldrick. Crystal structure refinement with SHELXL. *Acta Crystallogr., Sect. C: Struct. Chem.*, **2015**, 71(1), 3–8. <https://doi.org/10.1107/S2053229614024218>
19. M.R. Hansen, G.K.H. Madsen, H.J. Jakobsen, J. Skibsted. Refinement of borate structures from ¹¹B MAS NMR spectroscopy and density functional theory calculations of ¹¹B electric field gradients. *J. Phys. Chem. A*, **2005**, 109(9), 1989–1997 <https://doi.org/10.1021/jp045767i>
20. J.A. Bearden. X-ray wavelengths. *Rev. Mod. Phys.* **1967**, 39(1), 78–124. <https://doi.org/10.1103/RevModPhys.39.78>
21. P.J. Brown, A.G. Fox, E.N. Maslen, M.A. O'Keefe, B.T.M. Willis. Intensity of diffracted intensities. In: International Tables for Crystallography, Vol. C / Ed. E. Prince. Chester, England: International Union of Crystallography, **2006**, 554–595. <https://doi.org/10.1107/97809553602060000600>
22. A.L. Spek. What makes a crystal structure report valid? *Inorg. Chim. Acta*, **2018**, 470, 232–237. <https://doi.org/10.1016/j.ica.2017.04.036>

23. G. Viswanathan, A.O. Oliynyk, E. Antono, J. Ling, B. Meredig, J. Brgoch. Single-crystal automated refinement (SCAR): a data-driven method for determining inorganic structures. *Inorg. Chem.*, **2019**, 58(14), 9004–9015. <https://doi.org/10.1021/acs.inorgchem.9b00344>
24. O. Sologub, P. Rogl, A. Grytsiv, G. Giester. Crystal structure of $R_3Pd_{25-x}B_{8-y}$, $R = La, Ce$. *J. Phys.: Conf. Ser.*, **2009**, 176, 012007. <https://doi.org/10.1088/1742-6596/176/1/012007>
25. BAND2022. Amsterdam, The Netherlands: SCM, Theoretical Chemistry, Vrije Universiteit, <http://www.scm.com>.
26. E.S. Kadantsev, R. Klooster, P.L. De Boeij, T. Ziegler. The formulation and implementation of analytic energy gradients for periodic density functional calculations with STO/NAO Bloch basis set. *Mol. Phys.*, **2007**, 105(19–22), 2583–2596. <https://doi.org/10.1080/00268970701598063>
27. J.P. Perdew, Y. Wang. Accurate and simple analytic representation of the electron-gas correlation energy. *Phys. Rev. B*, **1992**, 45(23), 13244–13249. <https://doi.org/10.1103/PhysRevB.45.13244>
28. J.P. Perdew, K. Burke, M. Ernzerhof. Generalized gradient approximation made simple. *Phys. Rev. Lett.*, **1996**, 77(18), 3865–3868. <https://doi.org/10.1103/PhysRevLett.77.3865>
29. E. Van Lenthe, E.J. Baerends. Optimized slater-type basis sets for the elements 1–118. *J. Comput. Chem.*, **2003**, 24(9), 1142–1156. <https://doi.org/10.1002/jcc.10255>
30. P.H.T. Philipsen, E.J. Baerends. Relativistic calculations to assess the ability of the generalized gradient approximation to reproduce trends in cohesive properties of solids. *Phys. Rev. B*, **2000**, 61(3), 1773–1778. <https://doi.org/10.1103/PhysRevB.61.1773>
31. F.M. Bickelhaupt, E.J. Baerends. Kohn-Sham density functional theory: Predicting and understanding chemistry. In: *Reviews in Computational Chemistry*, Vol. 15 / Eds. K.B. Lipkowitz, D.B. Boyd. Hoboken, NJ, USA: John Wiley & Sons, Inc., **2007**, 1–86. <https://doi.org/10.1002/9780470125922.ch1>
32. K.M. Powderly, S. Guo, H.E. Mitchell Warden, L.T. Nguyen, R.J. Cava. Metastable β -NdCo₂B₂: a triclinic polymorph with magnetic ordering. *Chem. Mater.*, **2021**, 33(16), 6374–6382. <https://doi.org/10.1021/acs.chemmater.1c01545>
33. P. Salamakha, O. Sologub, C. Rizzoli, A.P. Gonçalves, M. Almeida. Rh₃B_{2-x}, new structure type of binary borides with triclinic symmetry. *J. Solid State Chem.*, **2004**, 177(11), 4237–4243. <https://doi.org/10.1016/j.jssc.2004.06.040>
34. L.E. Pangilinan, S. Hu, S.G. Hamilton, S.H. Tolbert, R.B. Kaner. Hardening effects in superhard transition-metal borides. *Acc. Mater. Res.*, **2022**, 3(1), 100–109. <https://doi.org/10.1021/accountsmr.1c00192>
35. S. Motojima, K. Sugiyama. Chemical vapour deposition of tantalum diboride. *J. Mater. Sci.*, **1979**, 14(12), 2859–2864. <https://doi.org/10.1007/BF00611466>
36. V.V. Lozanov, A.V. Utkin, N.I. Baklanova. Microhardness and thermal expansion of some binary and ternary boride phases of the Ca–Ir–B system. *Inorg. Mater.*, **2023**, 59(7), 696–702. <https://doi.org/10.1134/S0020168523070105>
37. R. Mohammadi, M. Xie, A.T. Lech, C.L. Turner, A. Kavner, S.H. Tolbert, R.B. Kaner. Toward inexpensive superhard materials: tungsten tetraboride-based solid solutions. *J. Am. Chem. Soc.*, **2012**, 134, 20660–20668. <https://doi.org/10.1021/ja308219r>
38. L.E. Pangilinan, C.L. Turner, G. Akopov, M. Anderson, R. Mohammadi, R.B. Kaner. Superhard tungsten diboride-based solid solutions. *Inorg. Chem.*, **2018**, 57, 15305–15313. <https://doi.org/10.1021/acs.inorgchem.8b02620>
39. A.T. Lech, C.L. Turner, J. Lei, R. Mohammadi, S.H. Tolbert, R.B. Kaner. Superhard rhenium/tungsten diboride solid solutions. *J. Am. Chem. Soc.*, **2016**, 138(43), 14398–14408. <https://doi.org/10.1021/jacs.6b08616>
40. G. Akopov, W.H. Mak, D. Koumoulis, H. Yin, B. Owens-Baird, M.T. Yeung, M.H. Muni, S. Lee, I. Roh, Z.C. Sobell, P.L. Diaconescu, R. Mohammadi, K. Kovnir, R.B. Kaner. Synthesis and characterization of single-phase metal dodecaboride solid solutions: Zr_{1-x}Y_xB₁₂ and Zr_{1-x}U_xB₁₂. *J. Am. Chem. Soc.*, **2019**, 141, 9047–9062. <https://doi.org/10.1021/jacs.9b03482>
41. R. Mohammadi, C.L. Turner, M. Xie, M.T. Yeung, A.T. Lech, S.H. Tolbert, R.B. Kaner. Enhancing the hardness of superhard transition-metal borides: Molybdenum-doped tungsten tetraboride. *Chem. Mater.*, **2016**, 28(2), 632–637. <https://doi.org/10.1021/acs.chemmater.5b04410>
42. M.T. Yeung, J. Lei, R. Mohammadi, C.L. Turner, Y. Wang, S.H. Tolbert, R.B. Kaner. Superhard monoborides: Hardness enhancement through alloying in W_{1-x}Ta_xB. *Adv. Mater.*, **2016**, 28(32), 6993–6998. <https://doi.org/10.1002/adma.201601187>
43. G. Akopov, L.E. Pangilinan, R. Mohammadi, R.B. Kaner. Perspective: superhard metal borides: a look forward. *APL Mater.*, **2018**, 6, 070901. <https://doi.org/10.1063/1.5040763>
44. P. Paufler, T. Weber. On the determination of linear thermal expansion coefficients of triclinic crystals using X-ray diffraction. *Eur. J. Mineral.*, **1999**, 11(4), 721–730. <https://doi.org/10.1127/ejm/11/4/0721>

45. F.G. Keihn, E.J. Keplin. High-temperature thermal expansion of certain group IV and group V diborides. *J. Am. Ceram. Soc.*, **1967**, 50(2), 81–84. <https://doi.org/10.1111/j.1151-2916.1967.tb15044.x>
46. S. Okada, K. Kudou, I. Higashi, T. Lundström. Single crystals of TaB, Ta₅B₆, Ta₃B₄ and TaB₂, as obtained from high-temperature metal solutions, and their properties. *J. Cryst. Growth*, **1993**, 128(1–4), 1120–1124.

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